

Biosynthesis

Key Aspects

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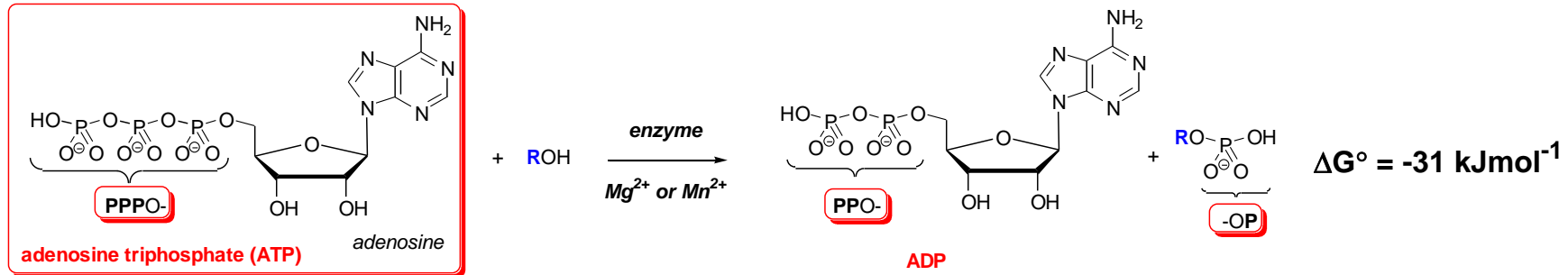
Dec 2014

Format & Scope of Lecture

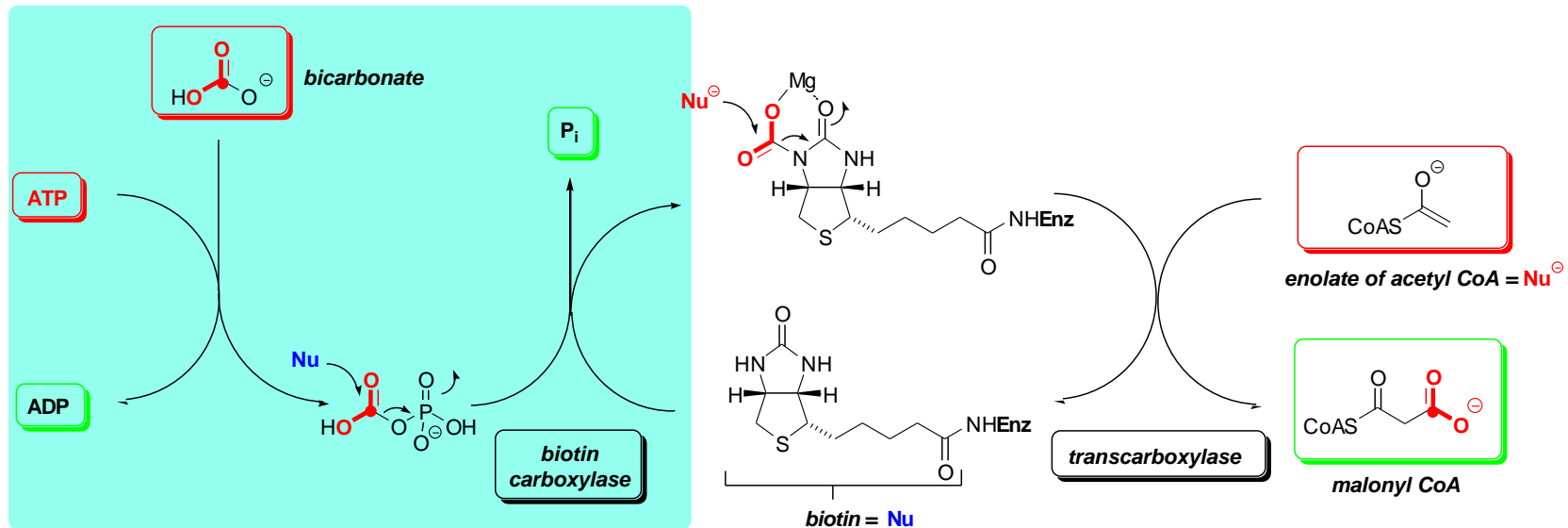
- **Enzyme cofactor chemistry:**
 - ATP, CoASH, SAM, DMAPP, biotin, NAD(P)⁺, NAD(P)H, FAD, FADH₂, peroxy-FADH₂, P₄₅₀ & PLP
- **Shikimate pathway:**
 - PEP + E-4-P → aromatic α-amino acids
- **Alkaloids**
 - Lys & Orn pathways to pyrrolidine & piperidine alkaloids – PLP chemistry
 - Phenolic coupling
- **Fatty acids and polyketides:**
 - The fatty acid synthase (FAS) iterative cycle
 - The polyketide synthase (PKS) iterative cycle(s)
- **Isoprenoids:**
 - The mevalonate pathway: 3x acetyl CoA → IPP & DMAPP
 - Pathways to linear C10, C15 & C20 isoprenoids: geranyl PP, farnesyl PP, geranylgeranyl PP
 - C30 isoprenoids *via* squalene synthase C15 head-to-head dimerisation
 - Basic carbocation chemistry: alkene cyclisation, elimination, trapping water, 1,2-alkyl & hydride shifts and modes of enzyme control thereof

ATP - Free Energy Releasing Couple

– Key process:

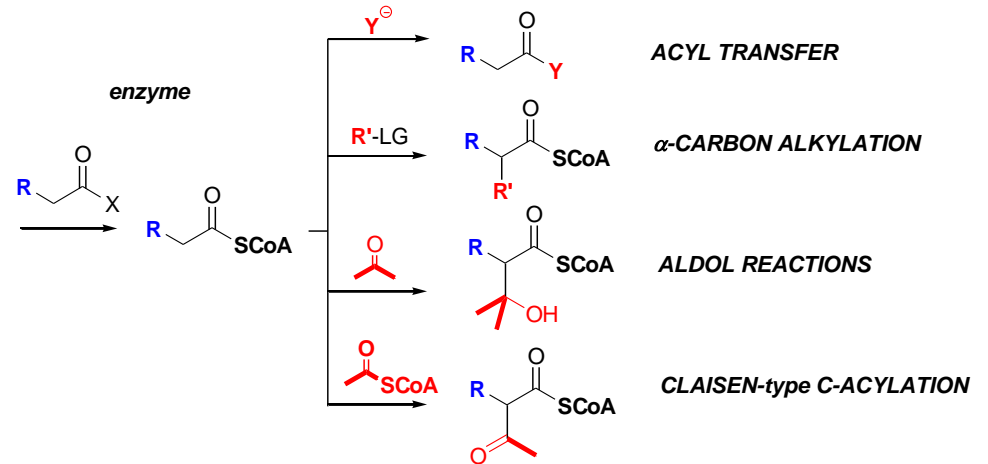
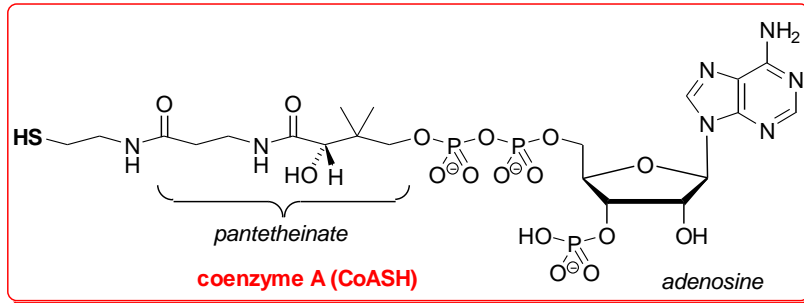


– e.g. activation of bicarbonate in malonyl CoA biosynthesis (FA + PK lectures)

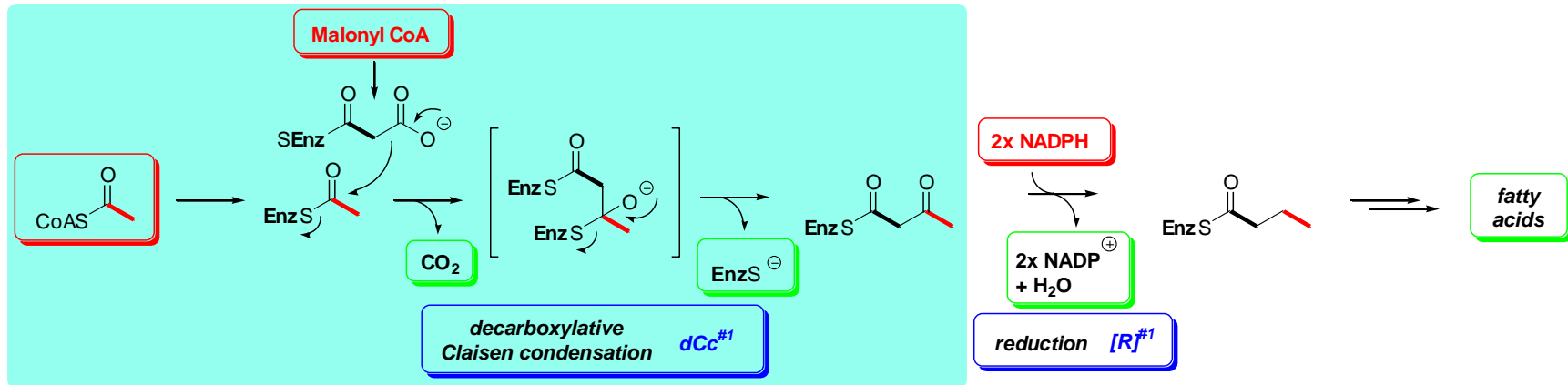


CoASH - C-C Bond Formation

- Key processes:



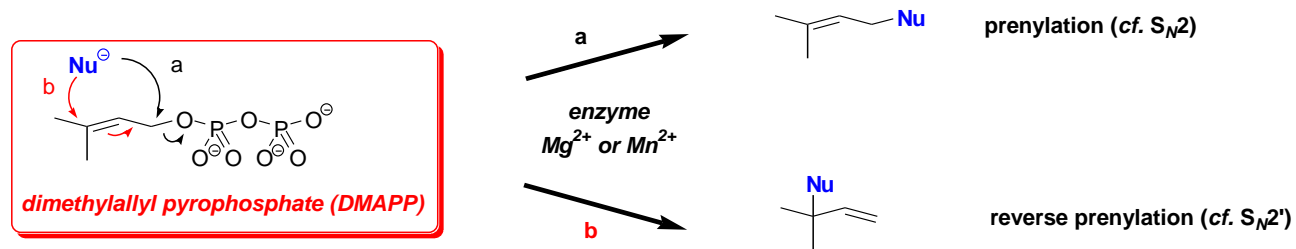
- e.g. iterative decarboxylative Claisen condensations in fatty acid biosynthesis (FA + PK lectures)



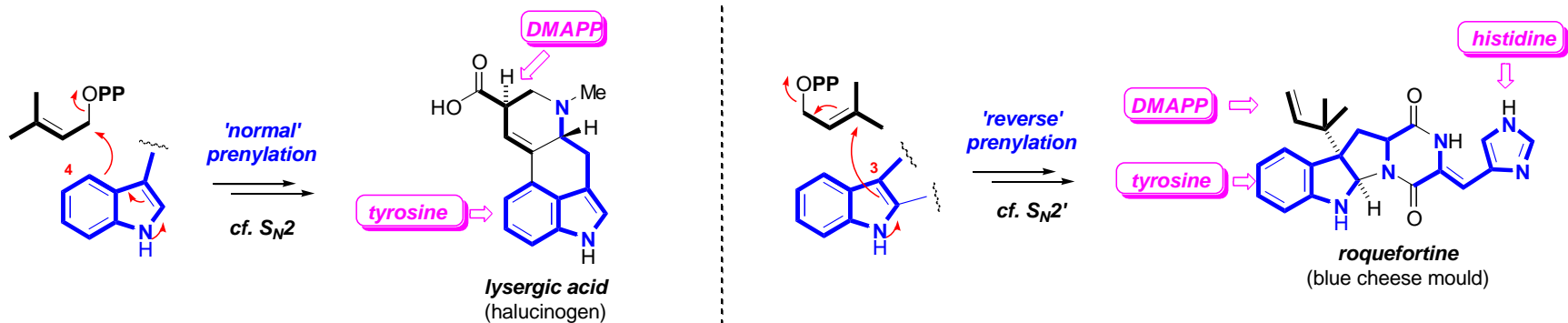
DMAPP - Dimethylallylation

– Key process:

- NB. via allylic carbocation with trapping by nucleophile at either most or least substituted end...

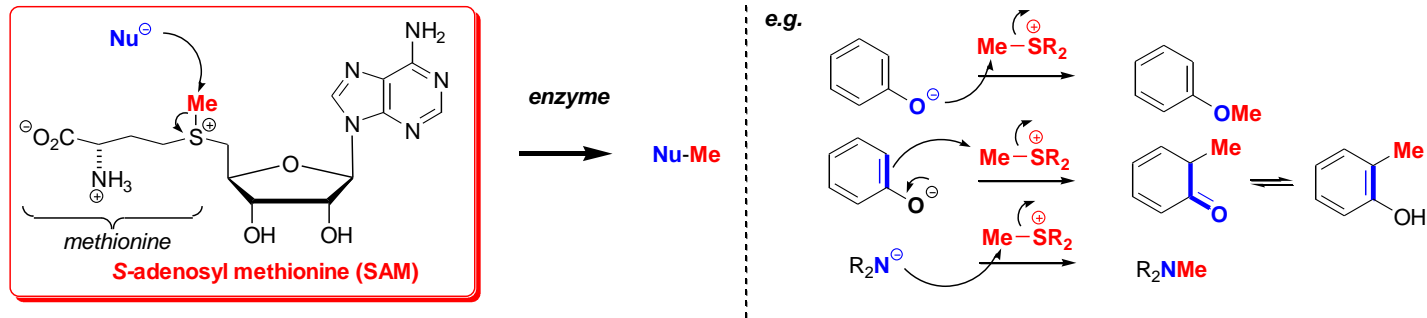


– e.g. prenylation & reverse prenylation of alkaloids (alkaloid lectures)

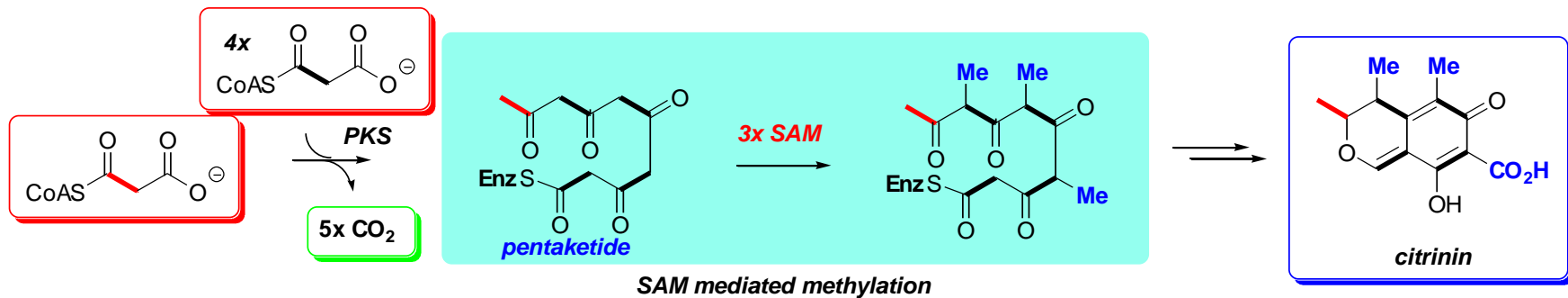


SAM - Methylation

- Key processes:

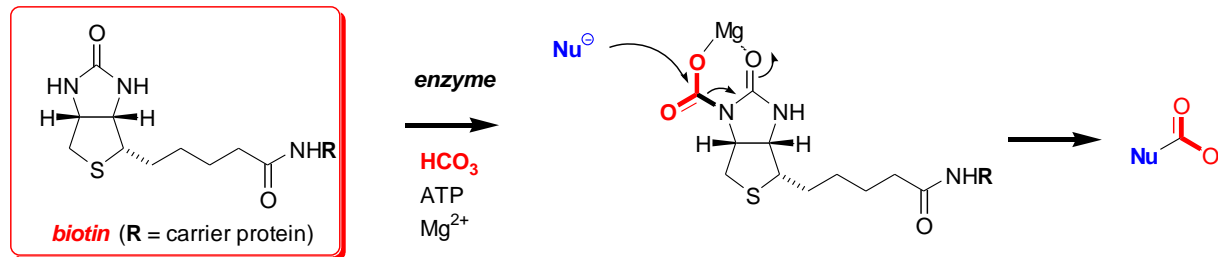


- e.g. ketone C-methylation in citrinin biosynthesis (FA + PK lectures)

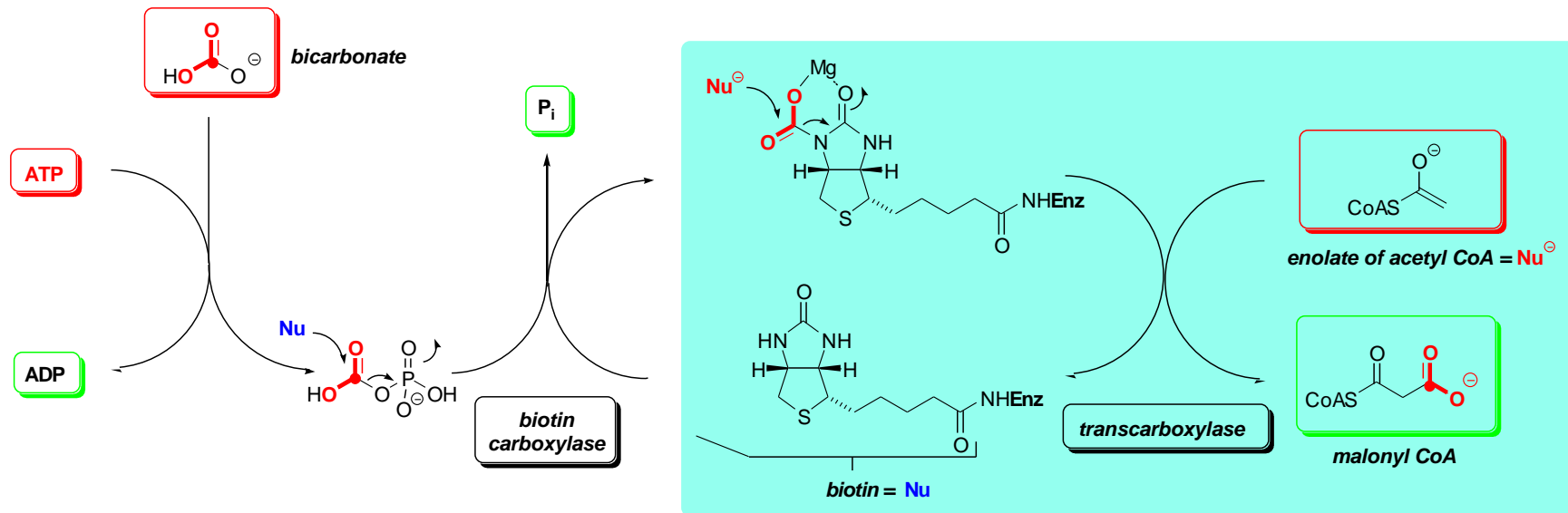


Biotin - Carboxylation

- Key process:

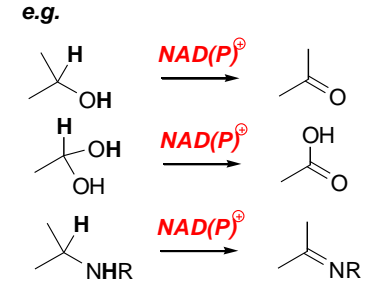
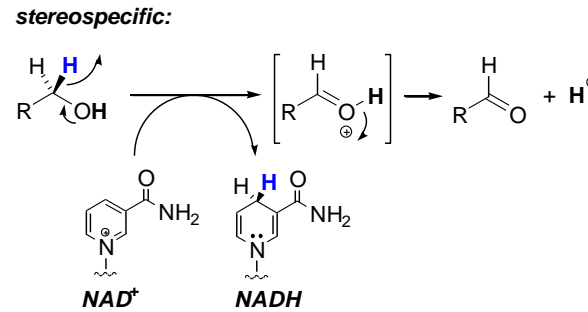
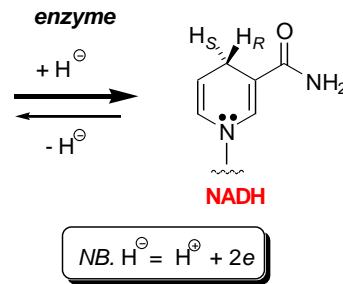
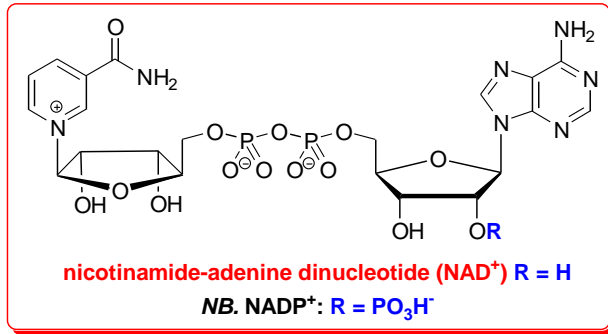


- e.g. carboxylation of acetyl CoA to give malonyl CoA (FA + PK lectures)

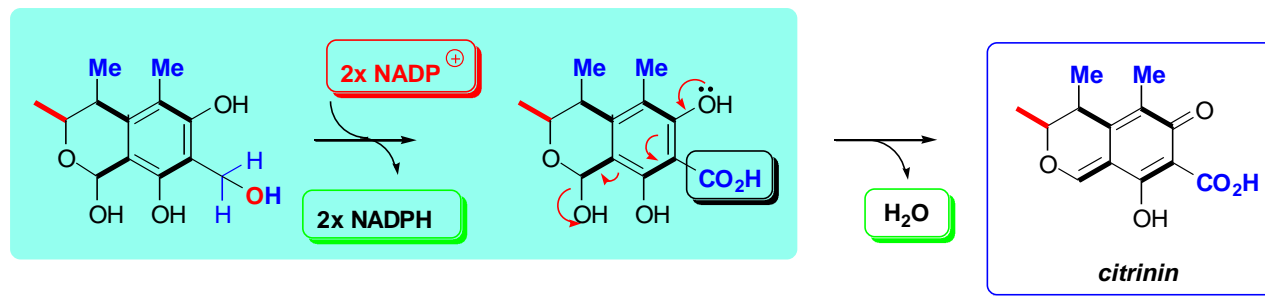


NAD(P)⁺ - Oxidation

– Key process:

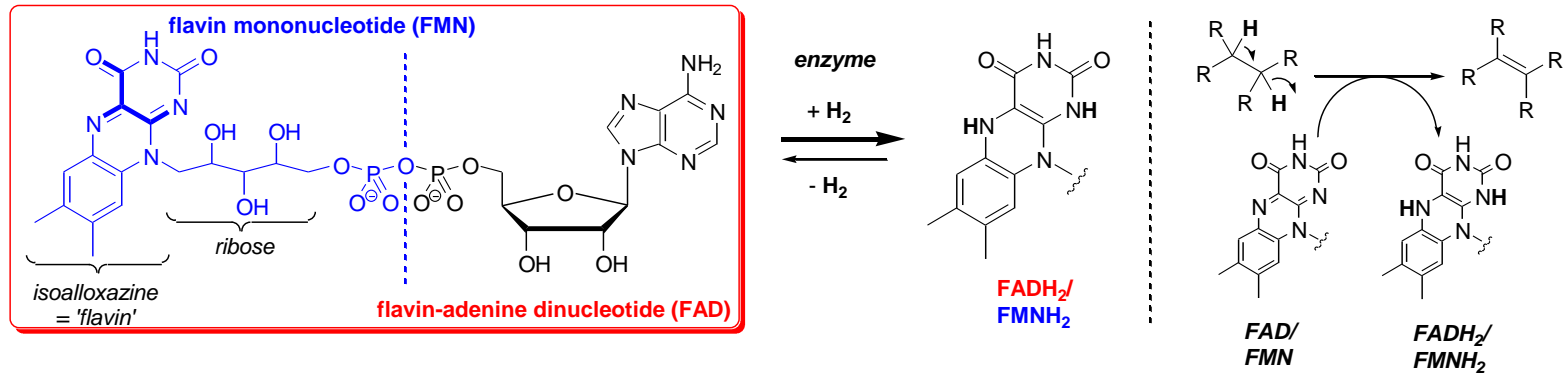


– e.g. 1° alcohol to carboxylic acid oxidation in citrinin biosynthesis (FA + PK lectures)

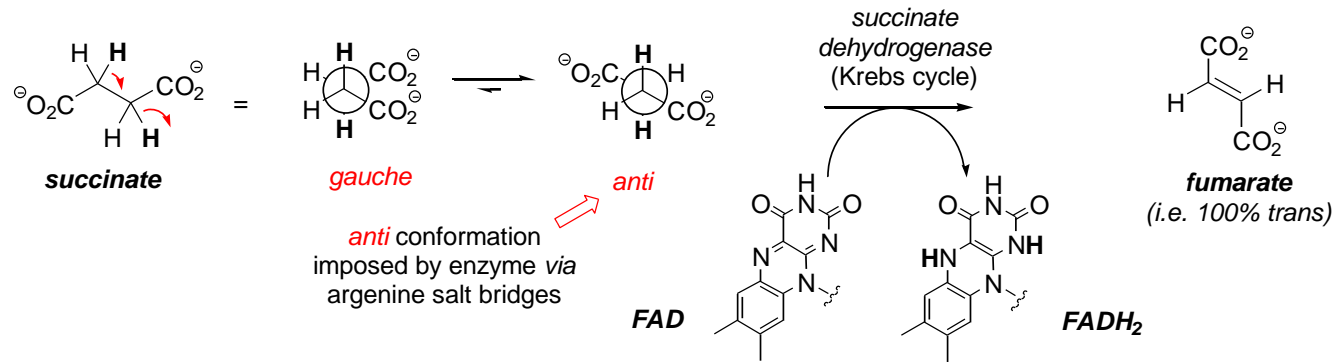


Flavin - Oxidation

- Key process:

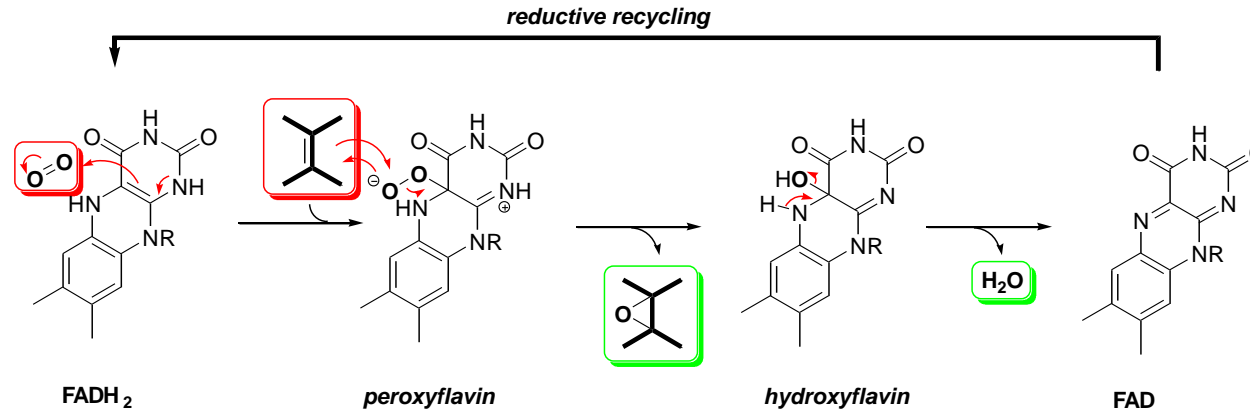


- e.g. dehydrogenation of succinate → fumarate (citric acid cycle, alkaloid lectures)

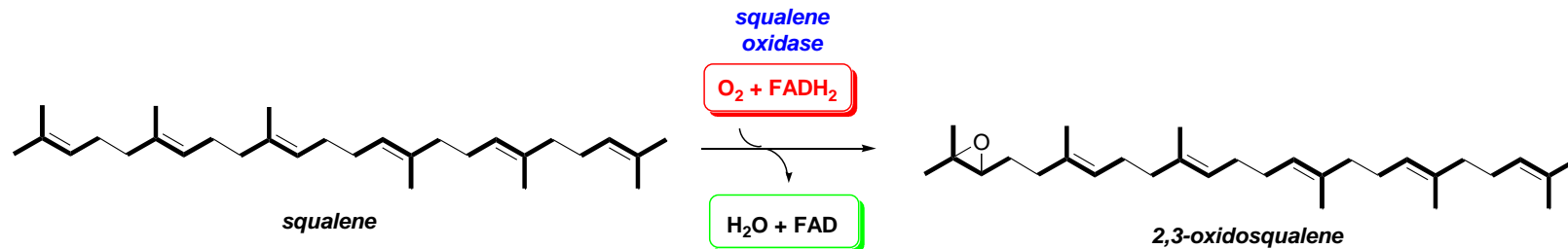


Peroxyflavin - Oxidation

- Key process:

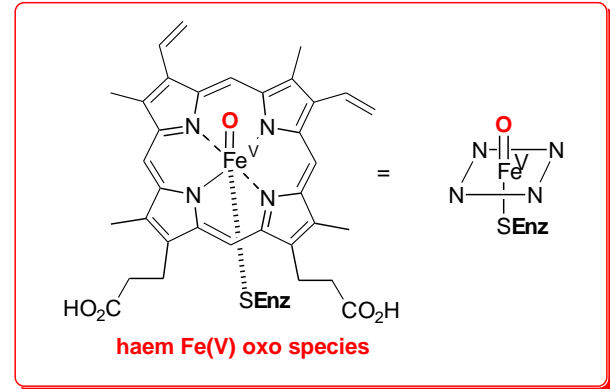
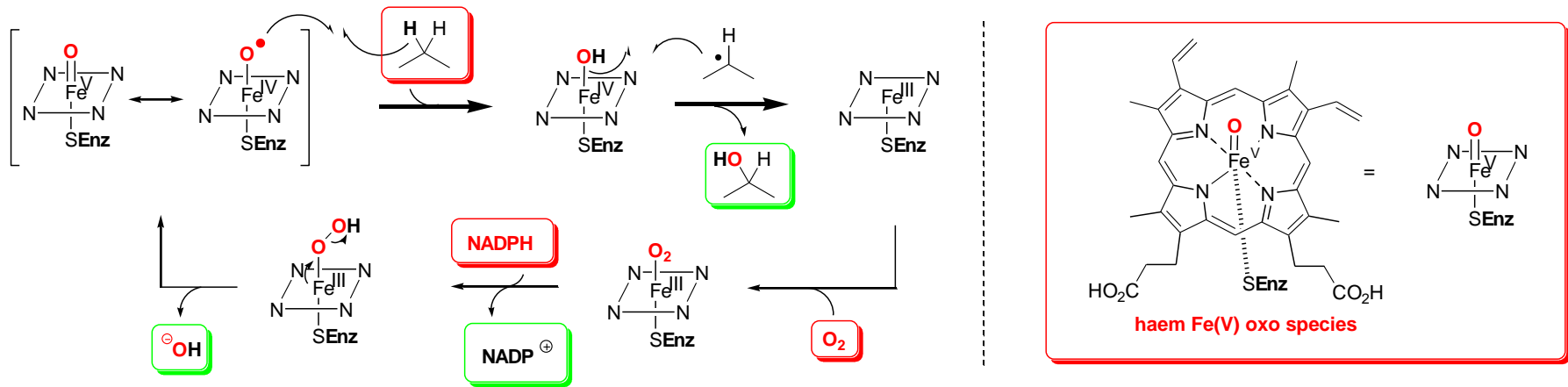


- e.g. squalene → 2,3-oxidosqualene (isoprenoid lectures)

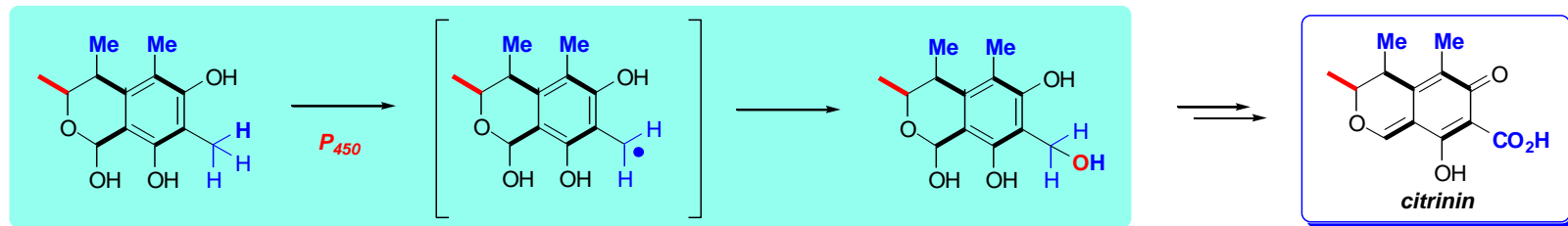


P_{450} Iron oxo - Oxidation

- Key process:



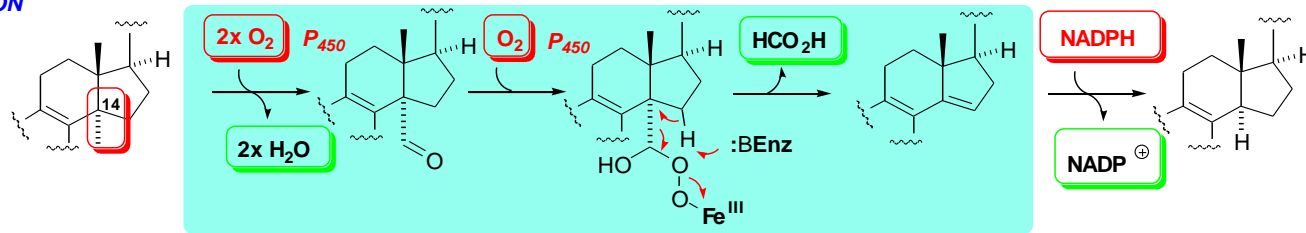
- e.g. 1) benzylic hydroxylation in citrinin biosynthesis (FA + PK lectures)



P_{450} Iron oxo – Oxidation cont.

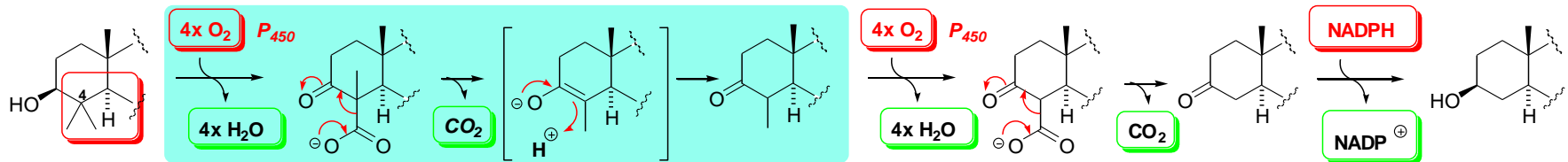
- e.g. 2) unactivated oxidative demethylation: 14α demethylation in steroid biosynthesis (isoprenoid lectures)

14α DEMETHYLATION



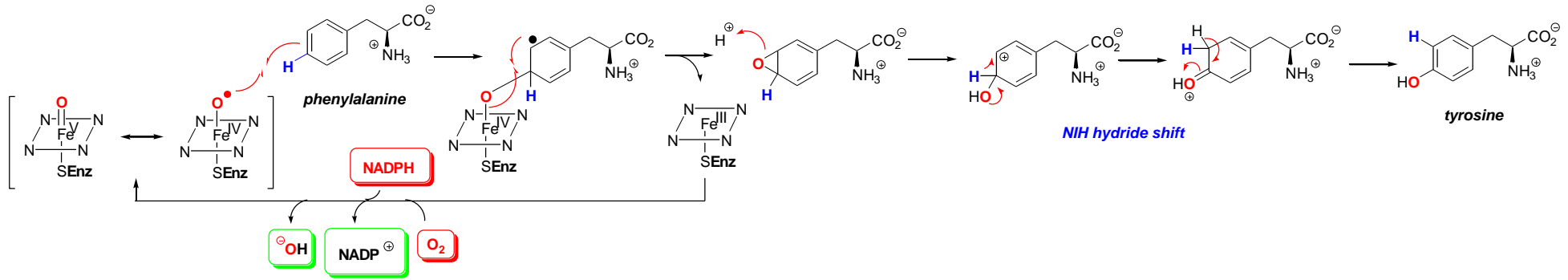
- e.g. 3) activated oxidative demethylation: $4\alpha/\beta$ demethylation in steroid biosynthesis (isoprenoid lectures)

4α & 4β DEMETHYLATION

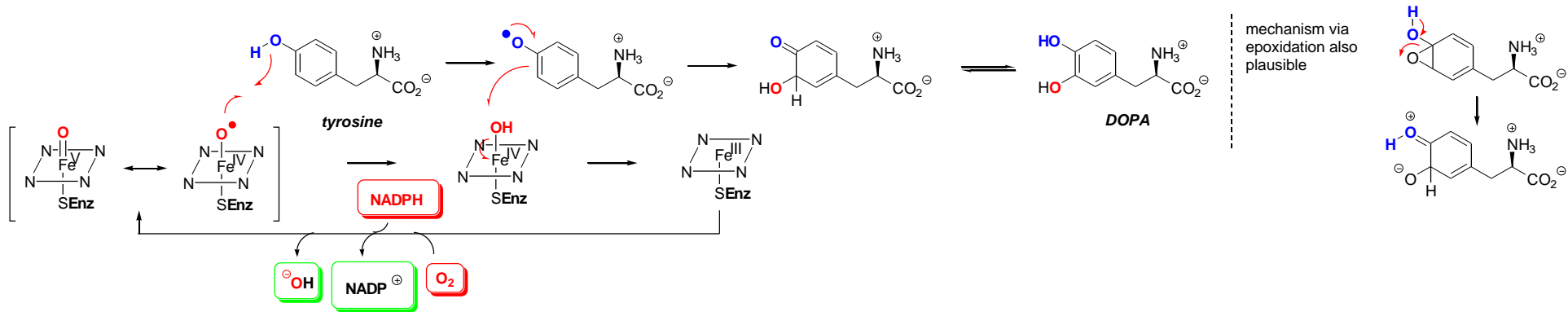


P_{450} Iron oxo – Oxidation cont.

- e.g. 4) unactivated aromatic hydroxylation: phenylalanine \rightarrow tyrosine in barley (alkaloid lectures)

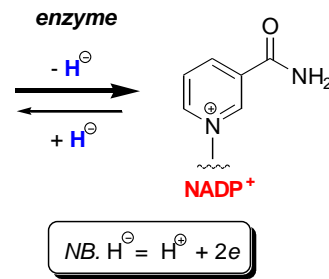
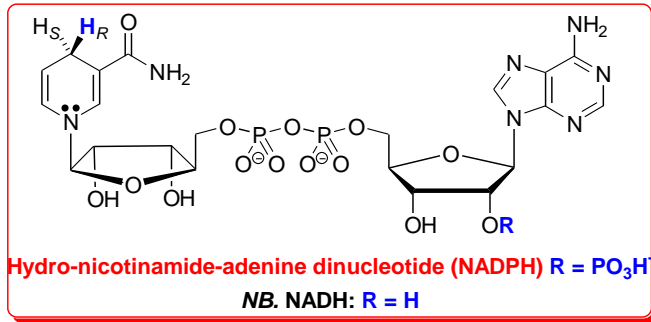


- e.g. 5) activated aromatic hydroxylation: tyrosine \rightarrow DOPA in opium (alkaloid lectures)

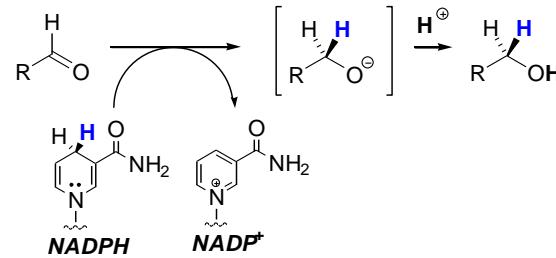


NADPH - Reduction

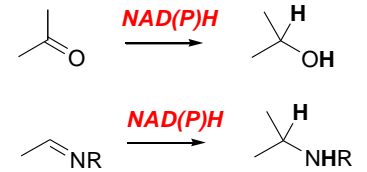
– Key process:



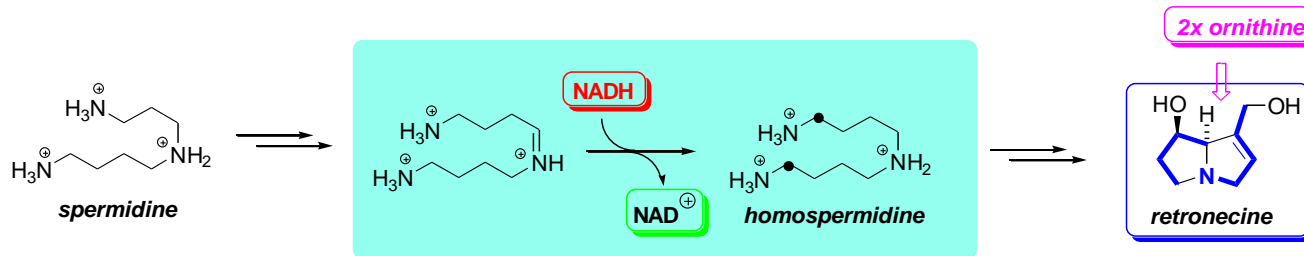
stereospecific:



e.g.

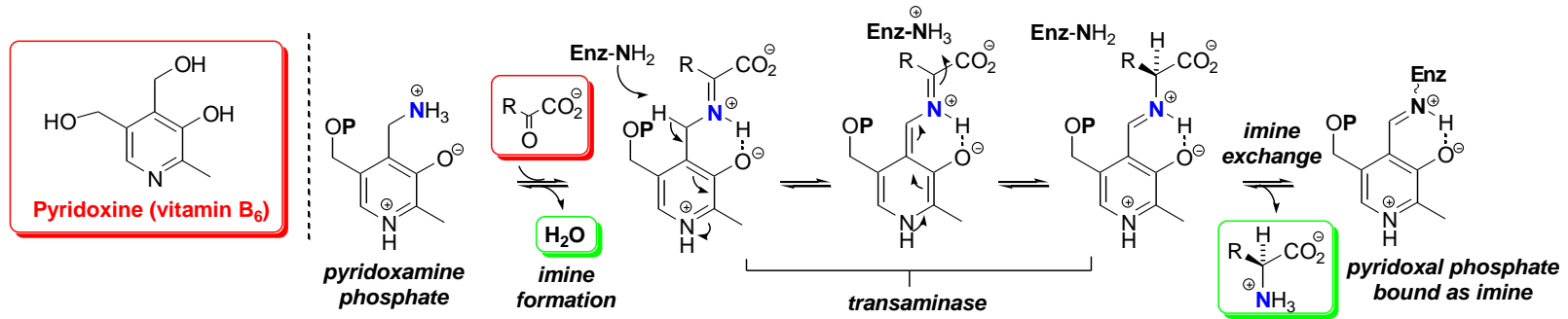


– e.g. iminium ion to amine reduction in pyrrolizidine alkaloid biosynthesis (alkaloid lectures)

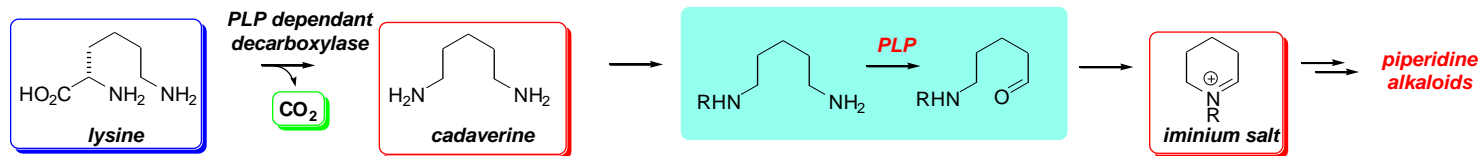


PLP - Transamination

- Key process:

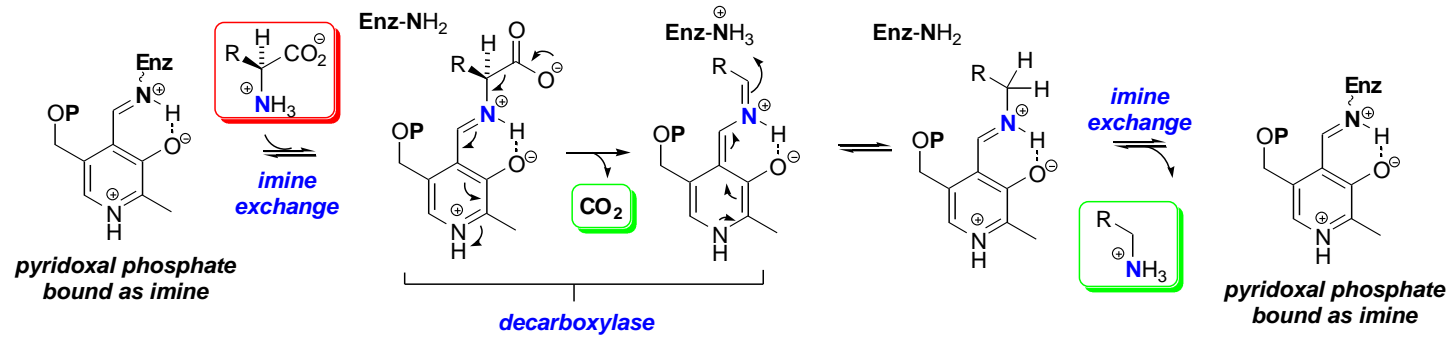


- e.g. oxidative deamination of cadaverine *en route* to piperidine alkaloids (alkaloid lectures)

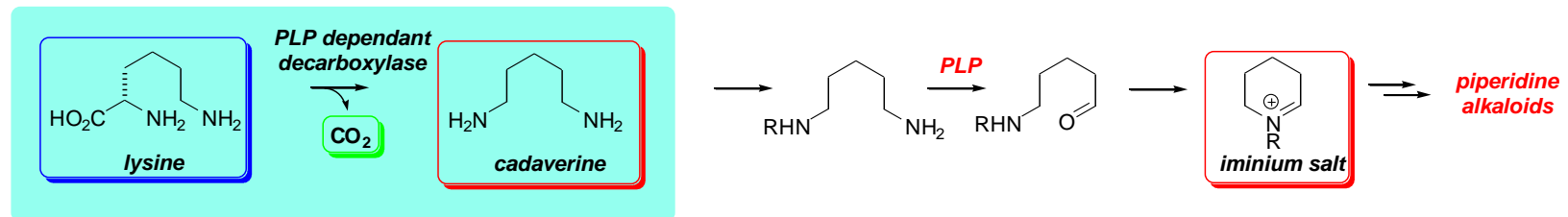


PLP - Decarboxylation

- Key process:

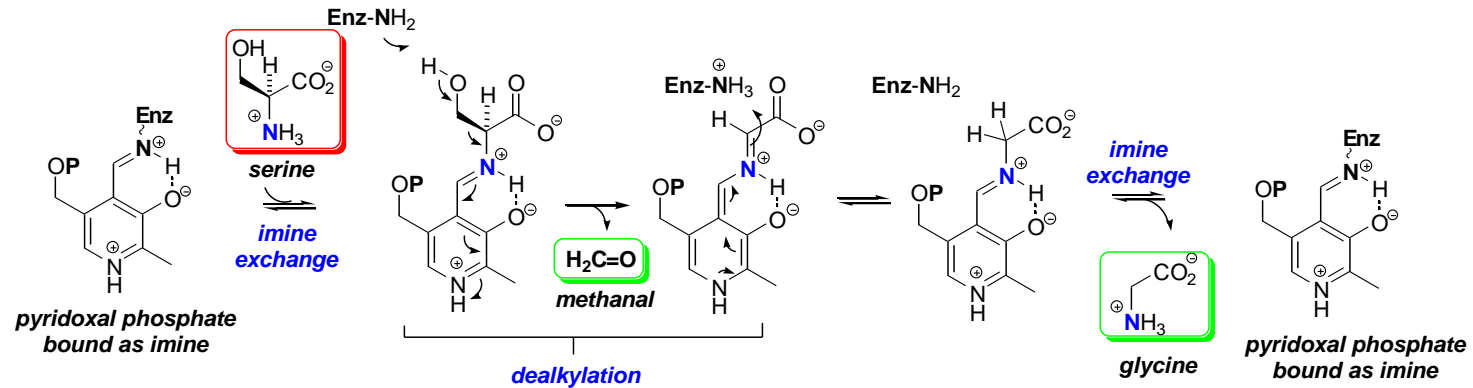


- e.g. lysine decarboxylation to cadaverine (alkaloid lectures)

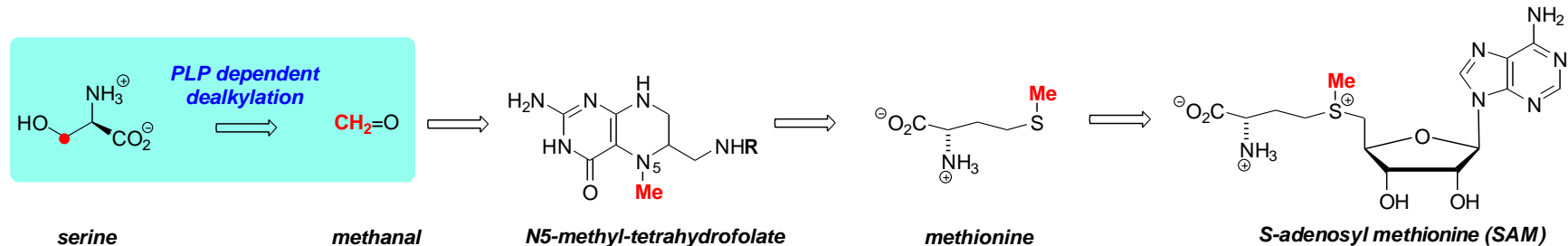


PLP - Dealkylation

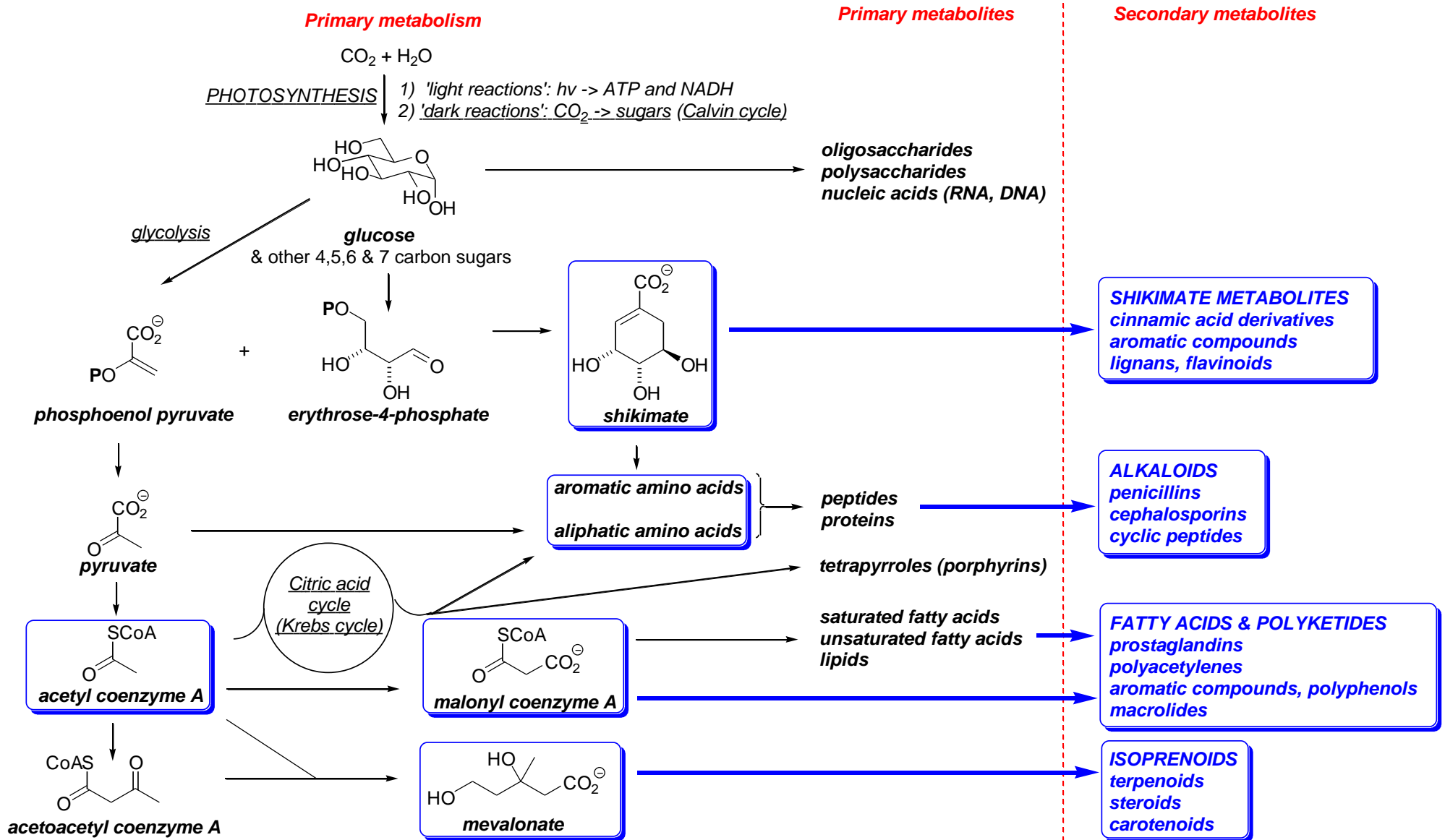
- Key process:



- e.g. serine side-chain cleavage \rightarrow methanal (alkaloid lectures)

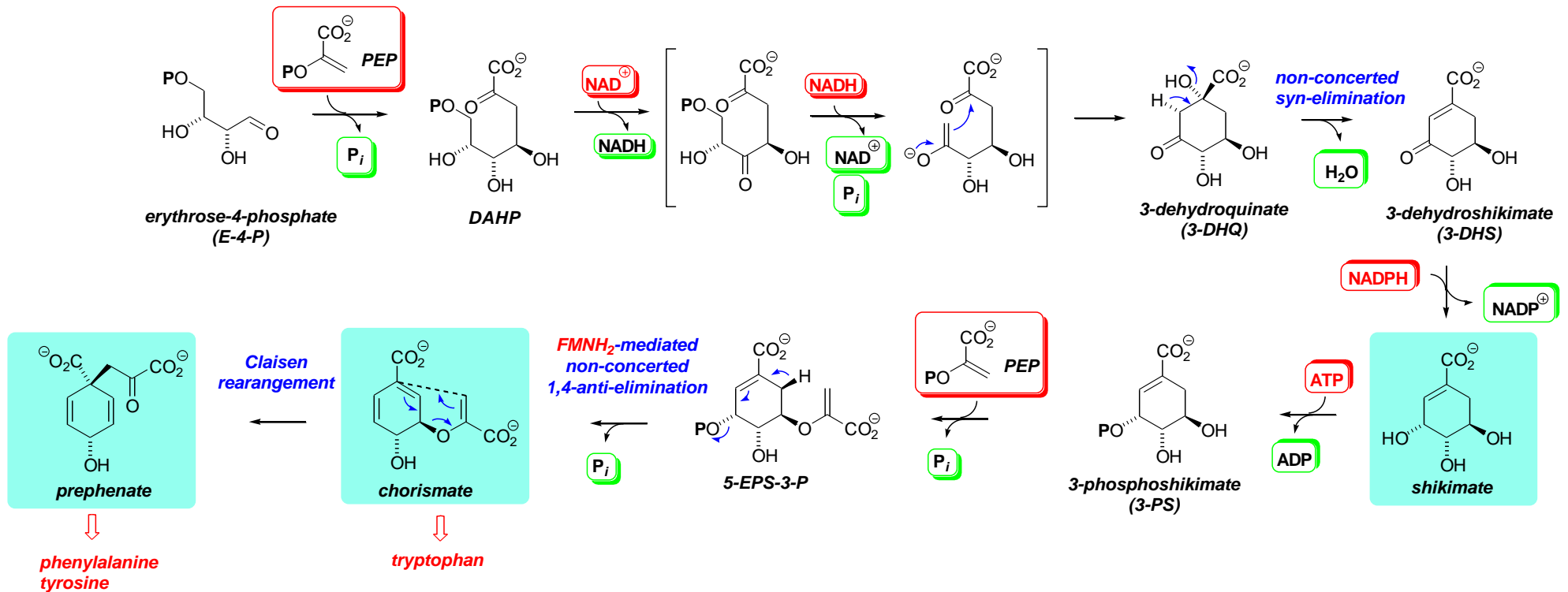


Primary Metabolism - Overview



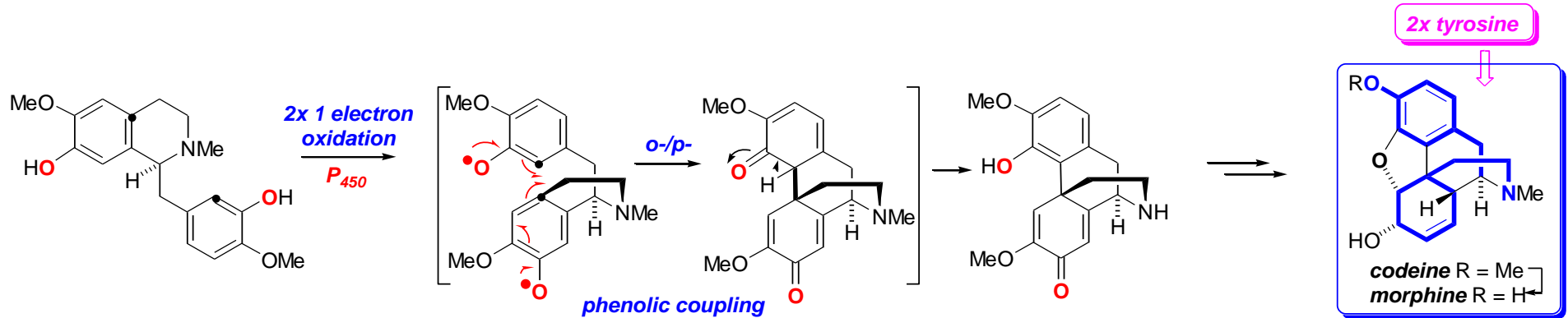
The Shikimate Biosynthetic Pathway

- Phosphoenol pyruvate & erythrose-4-phosphate → shikimate → chorismate → prephenate:

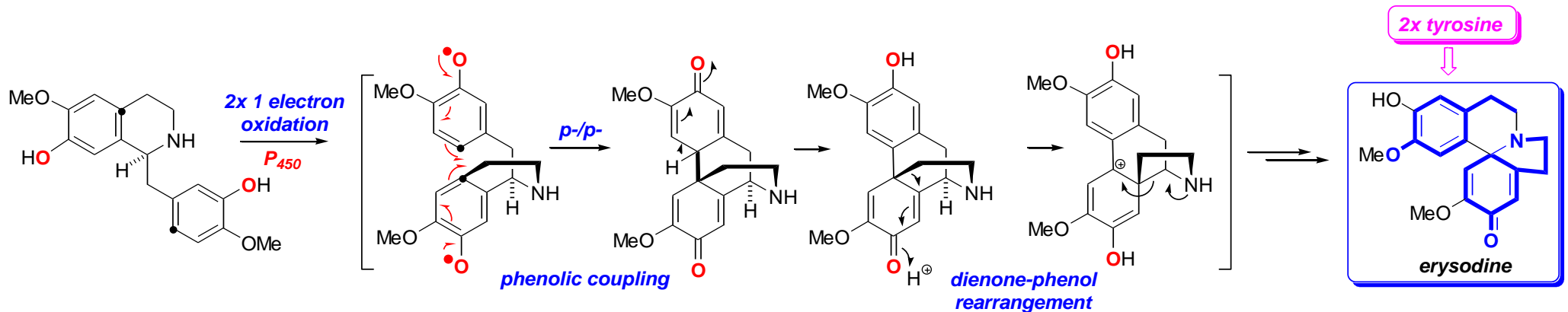


Oxidative Phenolic Coupling

- e.g. Morphine biosynthesis: o-/p- oxidative phenolic coupling:*

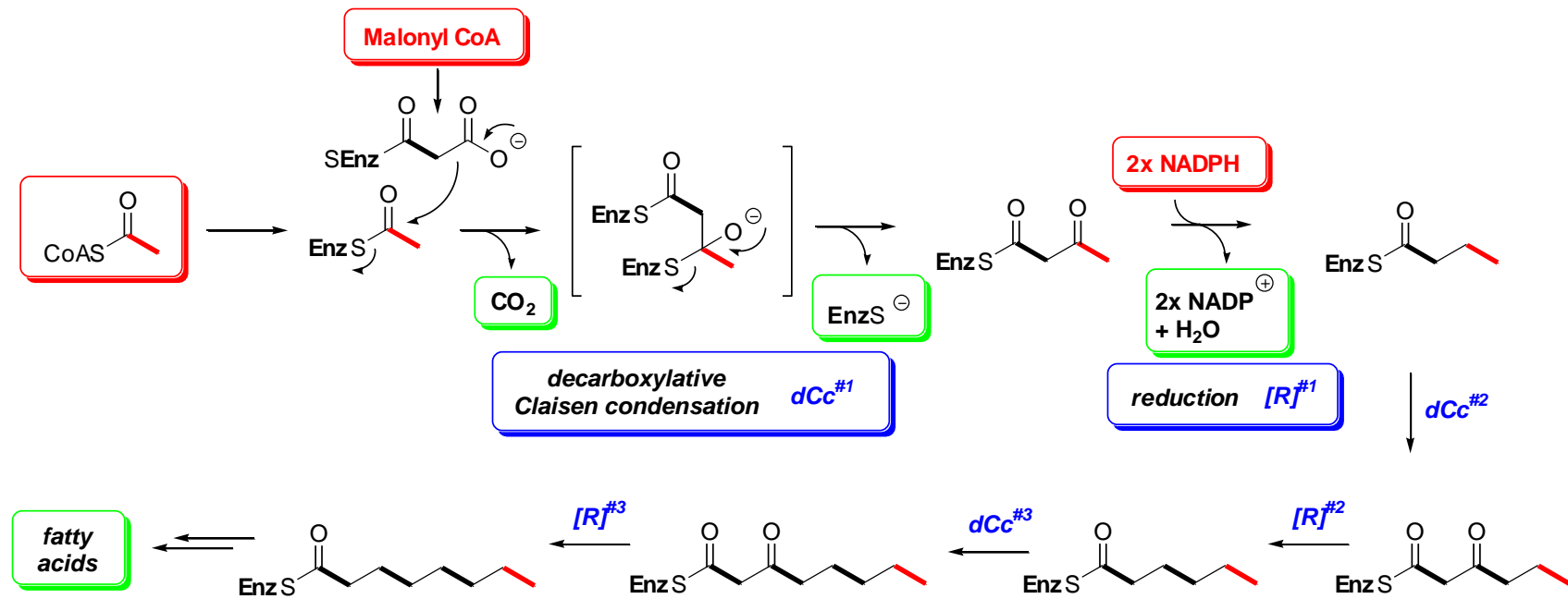


- e.g. Erysodine biosynthesis: p-/p- oxidative phenolic coupling & dienone-phenol rearrangement:*



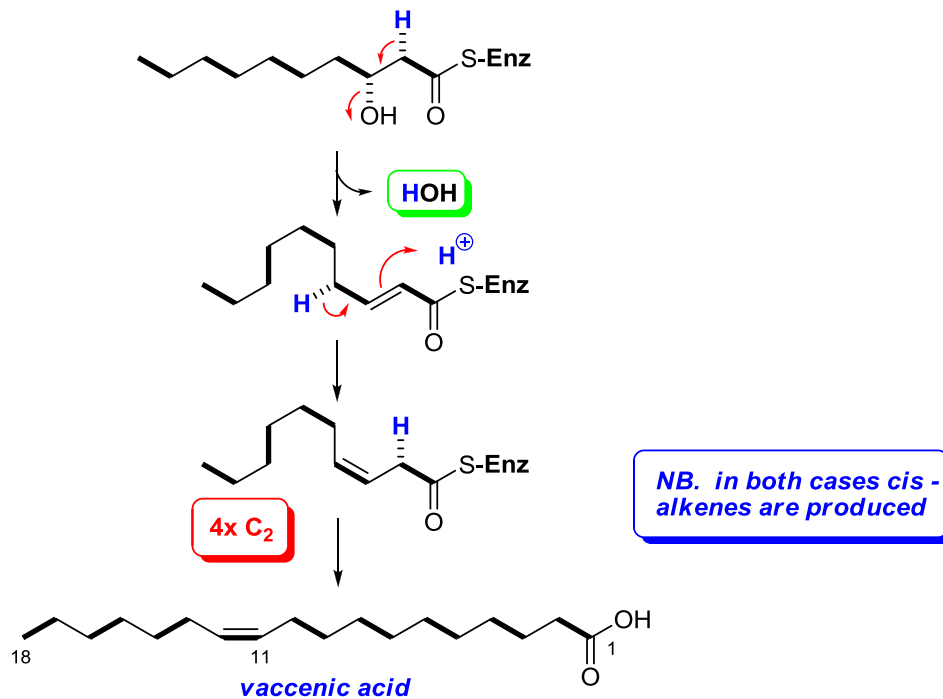
Biosynthesis of Fatty Acids

- Iterative oligomerisation *via*:
 - Decarboxylative Claisen condensation
 - 3-step ketone reduction

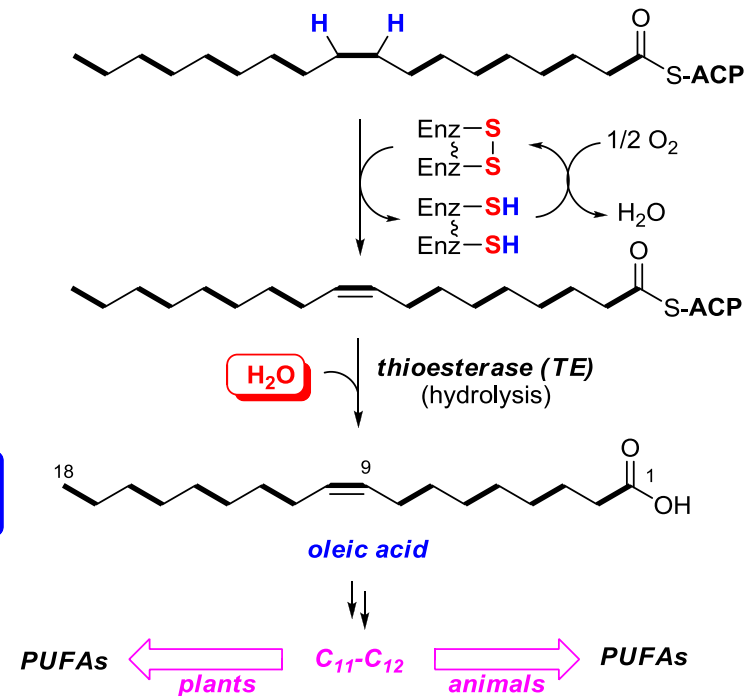


Introduction of Unsaturation

ANAEROBIC ROUTE (bacteria)
 (dehydrogenation occurs during chain elongation)
 mainly MUFAs but some PUFAs



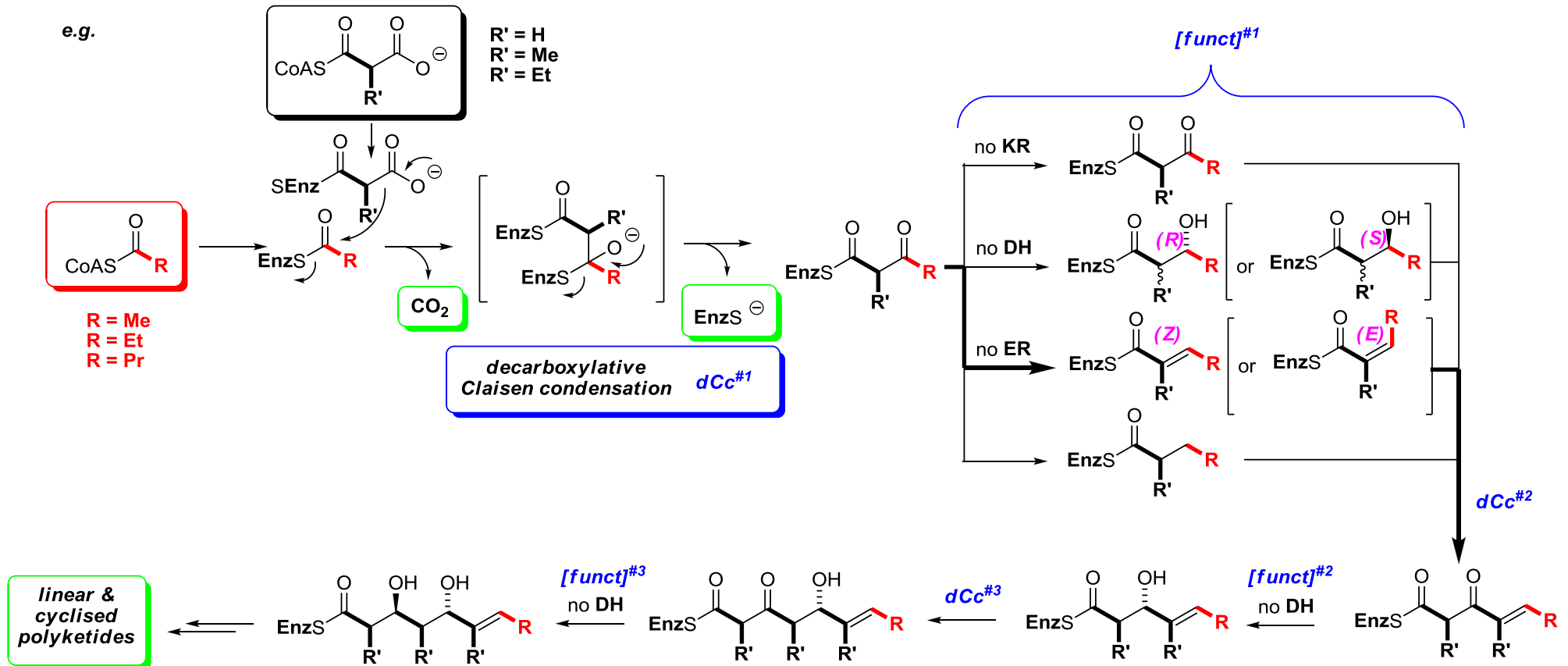
AEROBIC ROUTE (mammals, insects & plants)
 (dehydrogenation occurs after chain elongation)
 MUFAs & PUFAs



Biosynthesis of Polyketides

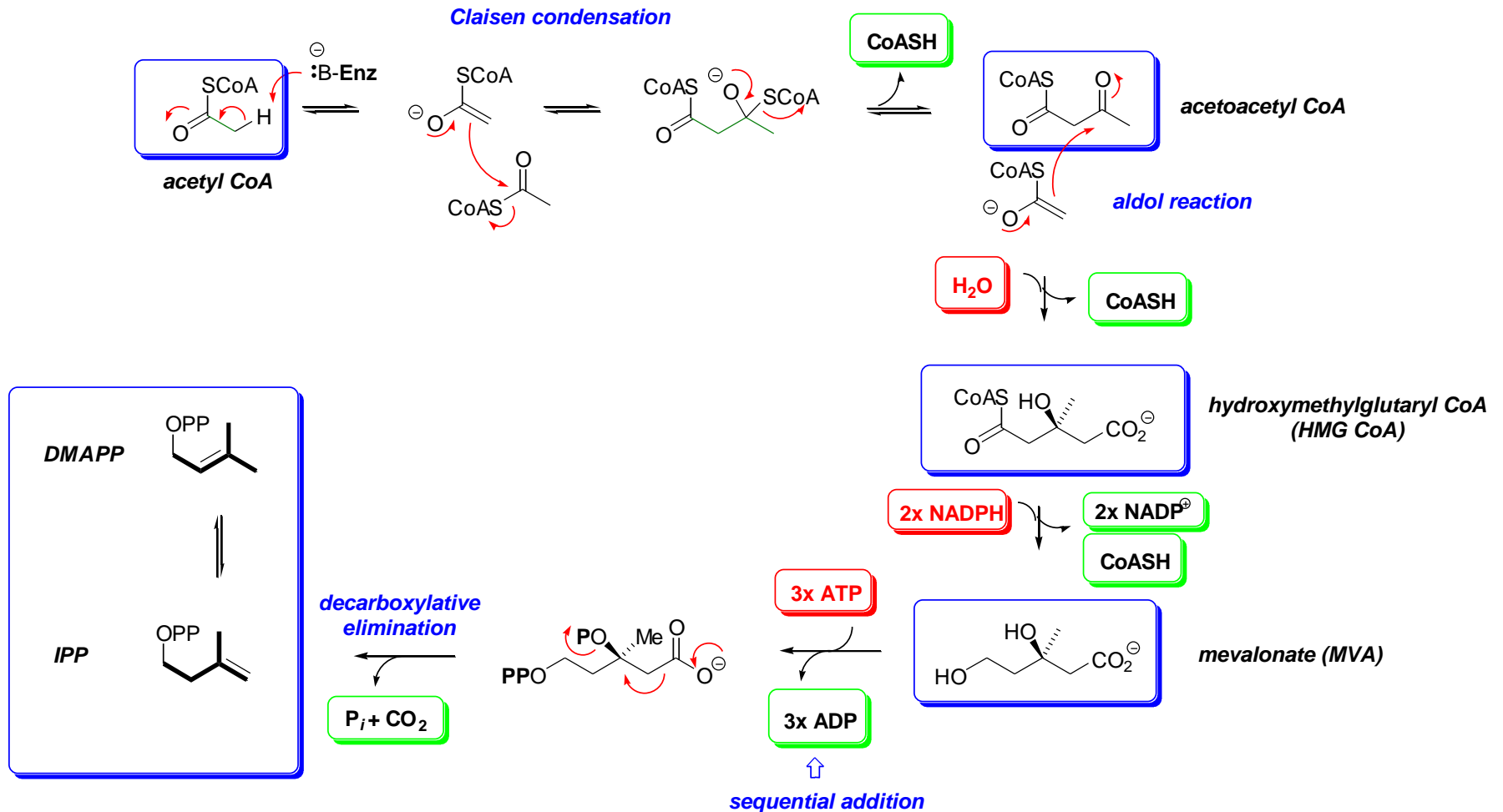
– Iterative oligomerisation *via*:

- Decarboxylative Claisen condensation
- Variable levels of reductive ketone processing in each iteration

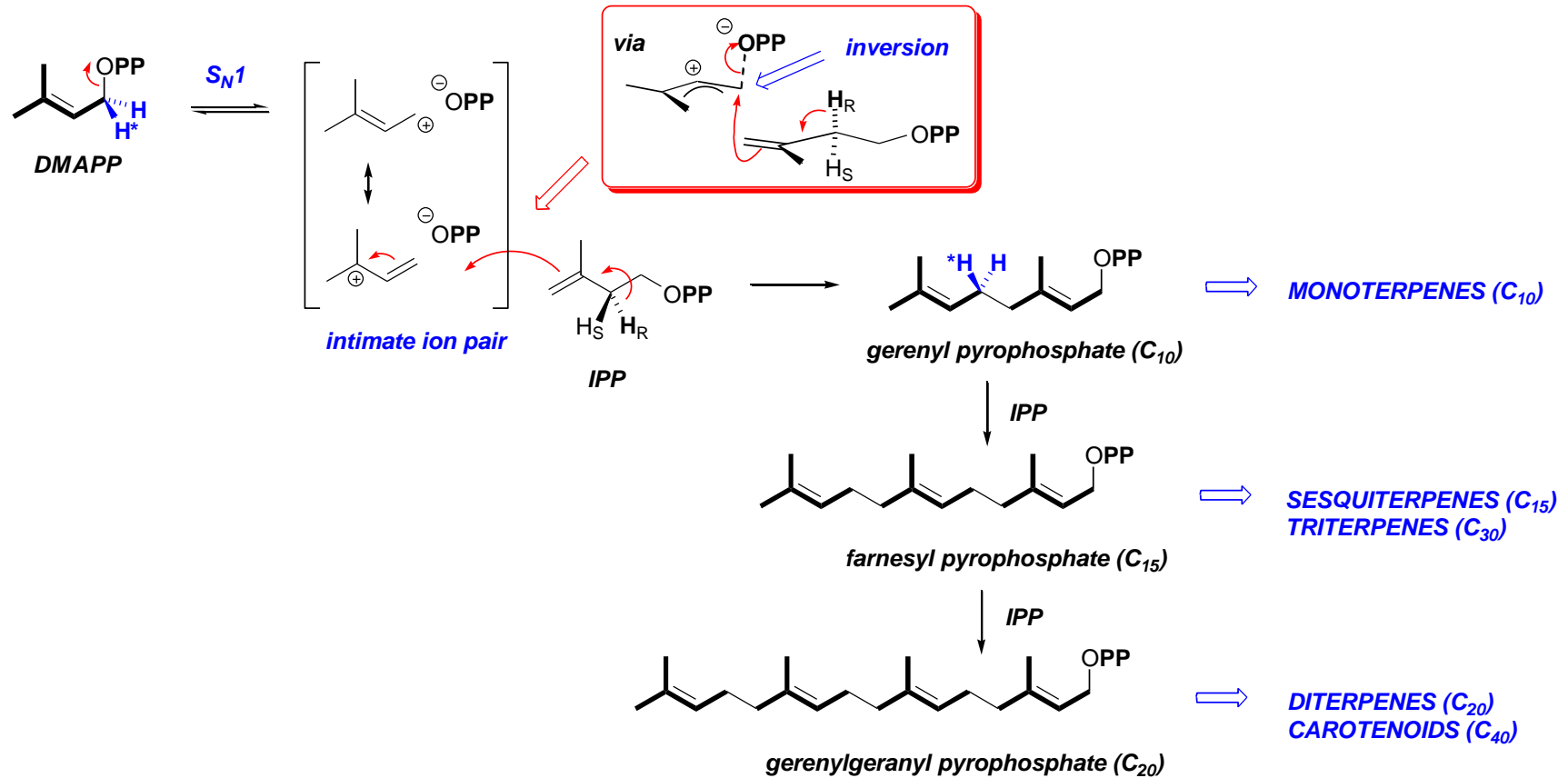


Biosynthesis of IPP & DMAPP

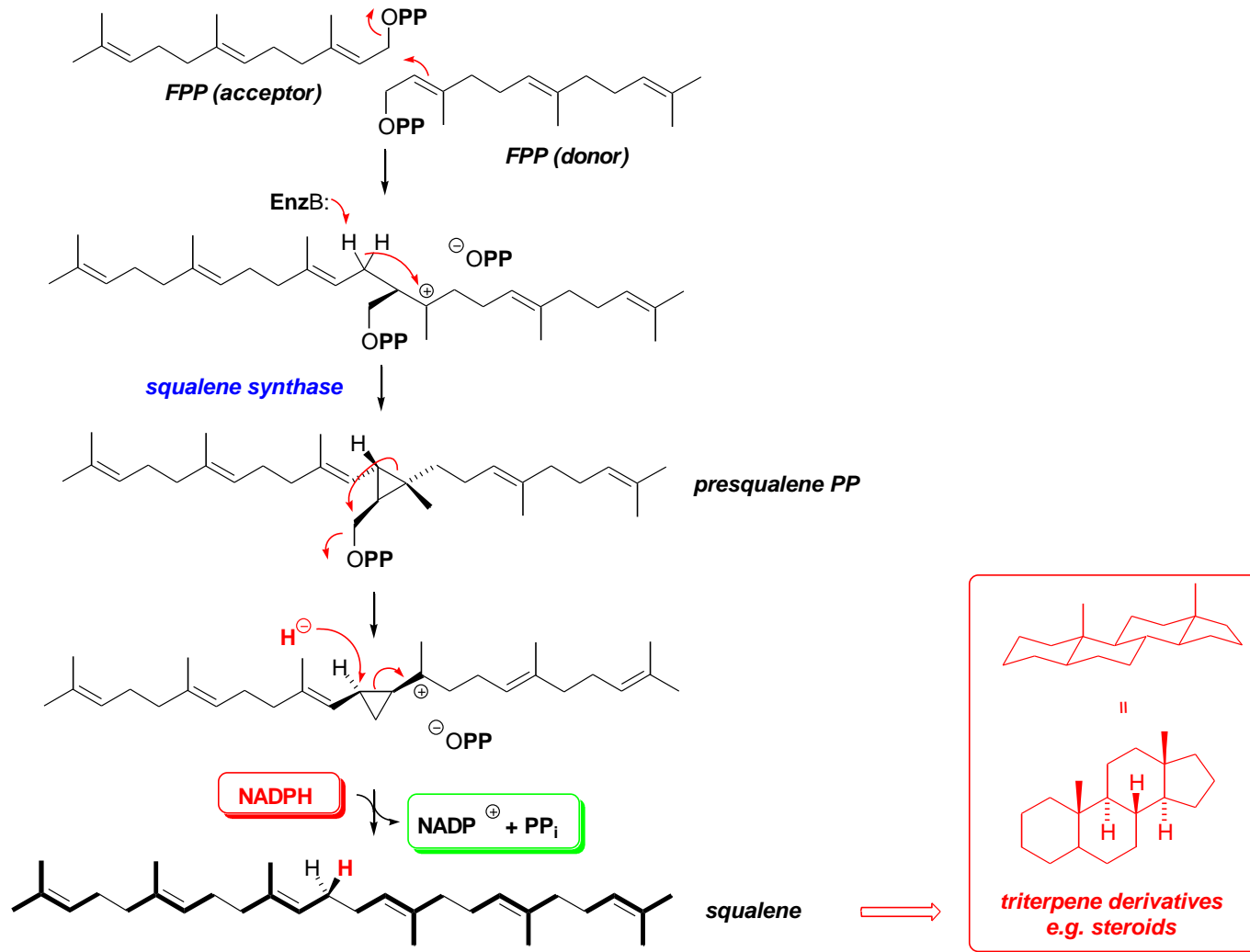
- acetyl CoA* → *acetoacetyl CoA* → *HMG CoA* → *mevalonate* → *IPP* → *DMAPP*:



'Head-to-tail' Oligomerisation → Isoprenoids



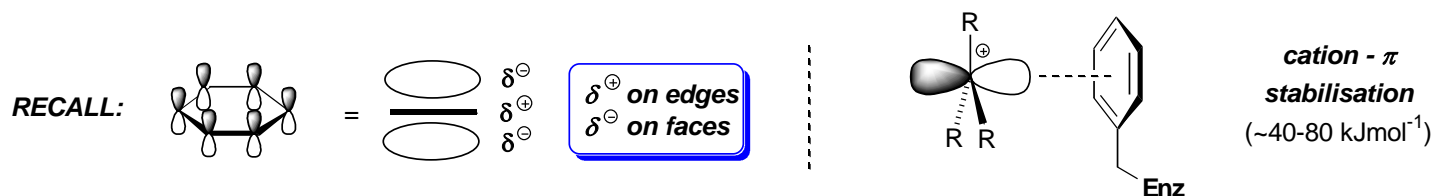
Triterpenes – 2x *FPP* 'head-to-head' → *Squalene*



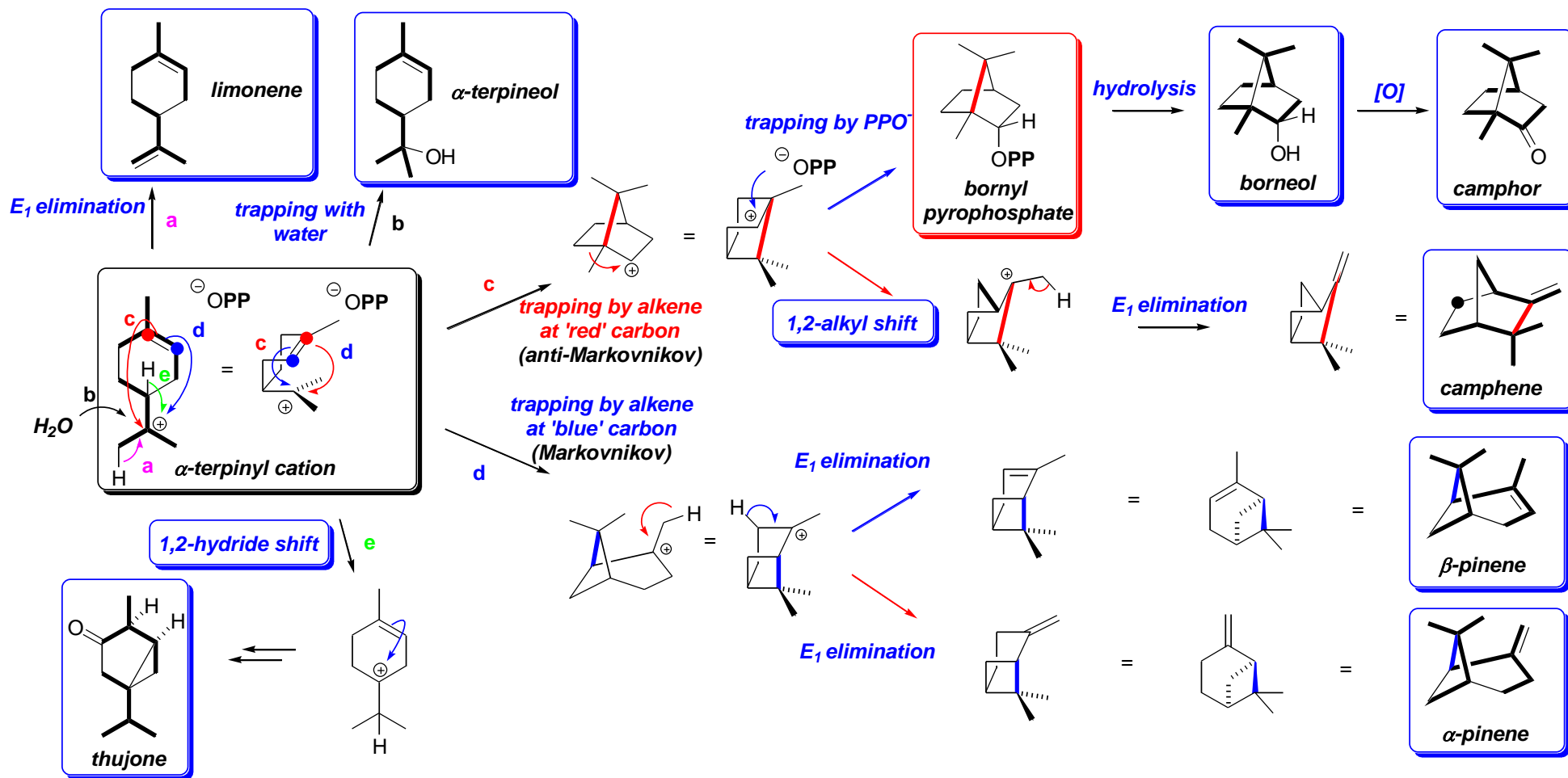
Terpene Cyclases – *Control of Cyclisation*

- **Functional aspects of terpenoid cyclases:**

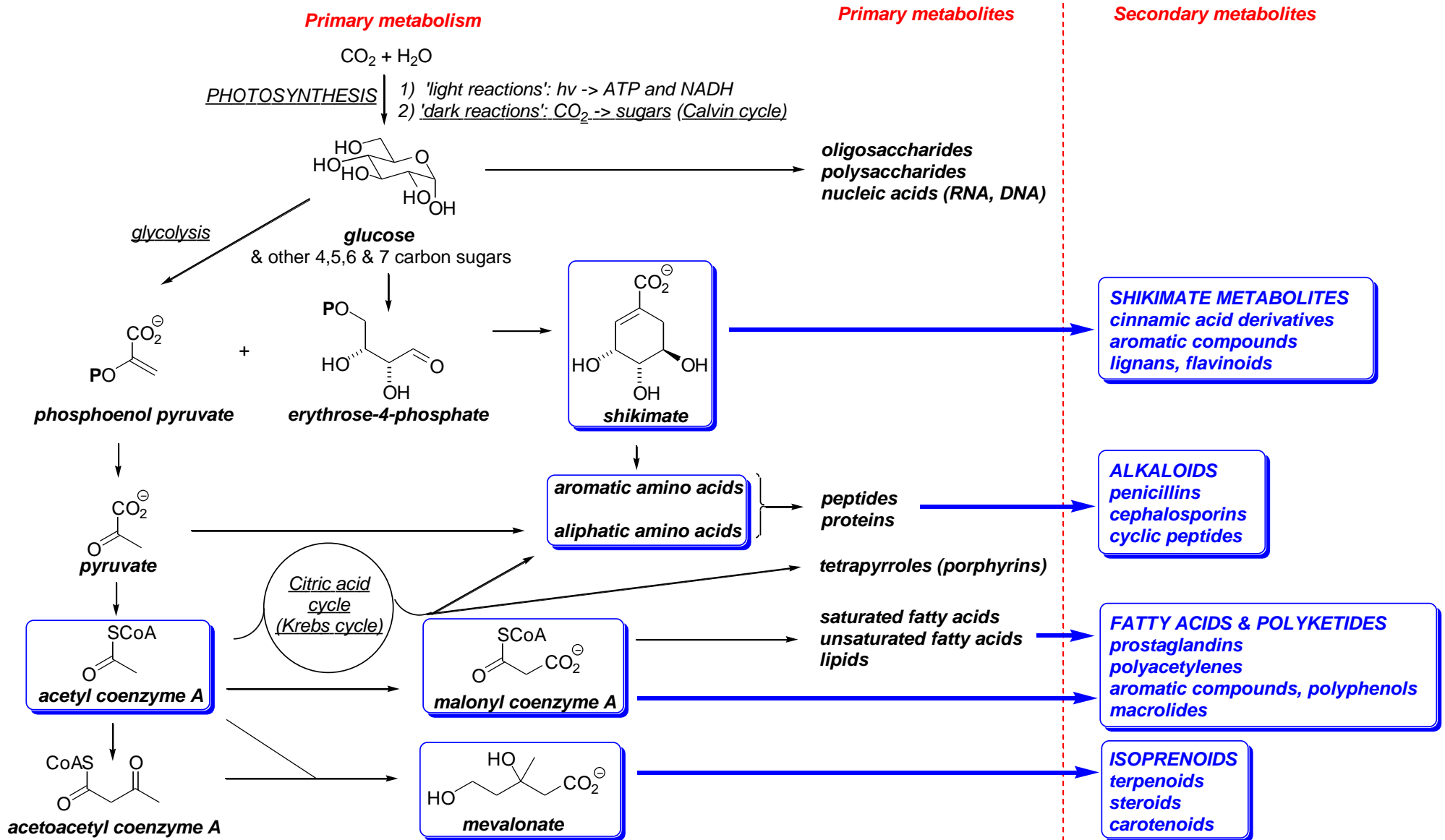
- **Templating:** Active site provides a template for a specific conformation of the flexible linear isoprenoid starting material.
- **Triggering:** Cyclase initiates carbocation formation.
 - Metal-assisted leaving group departure (e.g. pyrophosphate ionization aided by Mg^{2+})
 - C=C bond protonation (e.g. squalene-hopene cyclase, see later).
 - Epoxide protonation (e.g. oxidosqualene cyclase, see later).
- **Chaperoning:** Chaperones conformations of carbocationic intermediates through the reaction sequence, ordinarily leading to one specific product.
- **Sequestering:** Sequesters the carbocation intermediates by burying the substrate in a hydrophobic cavity that is generally solvent-inaccessible. Carbocations are concomitantly stabilized by the presence of aromatic residues in the active site that exert their effects *via* cation- π interactions



Cationic Rearrangements *etc.* → Diversity!



Primary Metabolism - Overview



Acknowledgements

- I would like to acknowledge the following for kindly allowing me to consult and use material from their lecture courses on various aspects of biosynthesis:



Dr Finian Leeper, Dept. of Chemistry, University of Cambridge

<http://www-leeper.ch.cam.ac.uk>



Dr John McKendrick, Dept. of Chemistry, University of Reading

<http://www.chem.rdg.ac.uk/dept/staff/org/jem.html>



Dr David Widdowson, Dept. of Chemistry, Imperial College London

<http://www.ch.ic.ac.uk/widdowson/>

- Additionally, I have adapted ideas from several web-sites & in particular I have adapted material from two biological chemistry courses at Harvard University & MIT:
 - <http://www.courses.fas.harvard.edu/%7echem27/>
 - <http://ocw.mit.edu/OcwWeb/Chemistry/5-08JSpring2004/CourseHome/index.htm>
- Other reference sources have been the books cited in the ‘course overview’, particularly:
 - **J. Mann**, ‘*Chemical Aspects of Biosynthesis*’, *Oxford Chemistry Primer No. 20*, 1994
 - **J. Mann**, ‘*Secondary Metabolism*’, *Oxford University Press*, 2nd ed. 1987